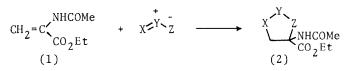
A SYNTHESIS OF GEMINALLY FUNCTIONALIZED HETEROCYCLIC AMINOCARBOXYLIC ACIDS BY CYCLOADDITION OF ETHYL N-ACETYL- α , β -DEHYDROALANINATE WITH 1,3-DIPOLES

Hiroshi Horikawa, Takashi Nishitani, Tameo Iwasaki, and Ichizo Inoue.

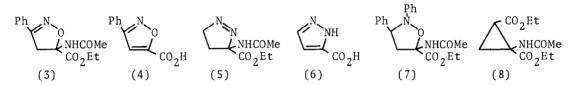
Department of Synthetic Chemistry, Research Laboratory of Applied Biochemistry, Tanabe Seiyaku Co., Ltd., Kashima-3-chome, Yodogawa-ku, Osaka 532, Japan

ABSTRACT 1,3-Dipolar cycloaddition of α , β -dehydroalaninate with dipoles proceeded with high regioselectivity to afford the title compounds in good yields.

 α,β -Dehydroalaninates are recognized to be intermediates of extensive value in both organic synthesis¹⁾ and biological transformations.²⁾ The vast majority of the literature dealing with the chemistry of α , β -dehydroalaninates involves nucleophilic and electrophilic addition reactions to give β and α substituted amino acids, respectively. Relatively few applications of the cycloaddition using α , β -dehydroalaninates to a synthesis of geminally functionalized cyclic aminocarboxylic acids have been described to date; some of these functionalized amino acids have been known³⁾ to have the noticeable biological activities. We have recently reported⁴⁾ that α,β -dehydroamino acids serve as a dienophile to afford the cycloaliphatic amino acids having an interesting carbon-skeleton. We now report that 1,3-dipolar cycloaddition of ethyl N-acetyl-a, B-dehydroalaninate (1) generally proceeds with remarkably high regioselectivity, thus providing a viable method for the synthesis of geminally funtionalized heterocyclic aminocarboxylic acids (2).



Ethyl N-acetyl- α , β -dehydroalaninate (1) was allowed to react with benzonitrile oxide in THF (r.t., 2hr) to give a single regioisomer, isoxazoline (3) (mp 136-137°C), in 85% yield: NMR (d_6 -DMSO) δ 1.21 (t, 3H), 1.90 (s, 3H), 3.77 (d, 2H), 4.18 (q, 2H), 7.2-7.8 (m, 5H), 9.23 (s, 1H). Support for the regiochemistry was obtained by treatment of (3) with hot dil.HCl which produced the known isoxazole (4) (mp 174-175°C; lit.,⁵⁾ mp 176-178°C) in 95% yield. (1) also underwent regiospecific 1,3-dipolar cycloaddition with diazomethane to afford pyrazoline (5) in 95% yield: NMR (CDC1₃) δ 1.27 (t, 3H), 2.07 (s, 3H), 1.9-2.3 (m, 2H), 4.25 (q, 2H), 4.6-5.9 (m, 2H), 7.38 (brs, 1H). In NMR spectra of (5), two methylene signals on the pyrazoline ring were observed separately as multiplets at the region of δ 2.0 and δ 5.0, indicating that (5) is the 3-disubsti-



tuted pyrazoline. The pyrazoline (5) was saponified, followed by acid treatment, to be converted into the known pyrazole (6) (mp 207-209°C: lit.⁶⁾ mp 216°C). The 1.3-dipolar cycloaddition was also carried out using C.N-diphenylnitrone as a dipole. Treatment of (1) with C,N-diphenylnitrone (50-60°C, 26hr, THF) provided a mixture of the stereoisomers⁷) of isoxazolidine (7) (mp 123-129°C) in 75% yield: NMR (CDC1_z) δ 1.22+1.28 (t+t, 3H), 1.81+1.99 (s+s, 3H), 2.6-3.0 (m, 2H), 4.58+4.95 (t+t, 1H), 4.18+4.24 (q+q, 2H), 6.7-7.2 (m, 5H), 7.2-7.6 (m, 5H), 7.97+ 8.20 (brs+brs, 1H). This cycloaddition has proved to proceed regiospecifically on the basis of the NMR spectra in which two methine triplets corresponding to the proton on C-3 were observed at δ 4.58 and δ 4.95. Thus, regardless of the presence of acetamido $group^{8}$; the regiochemistry and the rate of reaction in the cycloaddition using the α,β -dehydroalaninate show striking similarities to those in the case of ethyl acrylate.⁹⁾ This is further confirmed by the less reactivity of (1) with ethyl diazoacetate which easily reacts with electron rich dipolarophiles such as enamines¹⁰⁾: the reaction of (1) with ethyl diazoacetate was much slower than that with diazomethane, vielding the cyclopropane derivative (8),¹¹⁾ a single stereoisomer (mp 72-73°C), only in 12.5% yield (55°C, 24hr): NMR (CDC1₇) & 1.27 (t, 6H), 1.5-2.0 (m, 2H), 1.99 (s, 3H), 2.72 (t, 1H), 4.13 (q, 2H), 4.16 (q, 2H), 6.53 (s, 1H).

The application of the 1,3-dipolar cycloaddition with high regioselectivity to a preparation of biologically interesting functionalized amino acids is now under investigation.

References and Footnotes

References and Footnotes 1) R. K. Olsen and A. J. Kolar, Tatrahedron Lett., 3579 (1975) and references cited therein. 2) B. W. Bycroft, Nature, 224, 595 (1969) and references cited therein. 3) H. T. Nagasawa, J. A. Elberling, and F. N. Shirota, J. Med. Chem., 16, 823 (1973). 4) H. Horikawa, T. Nishitani, T. Iwasaki, Y. Mushika, I. Inoue, and M. Miyoshi, Tatrahedron Lett., 4101 (1980). 5) A. Quilico and G. Specroni, Gazz. Chim. Ital., 76, 148 (1946). 6) I. Ichimoto, K. Fujii, and C. Tatsumi, Agr. Biol. Chem., 31, 979 (1967). 7) The ratio was found to be 5:2 from the NMR assignment of the mixture, although the spectrum does not unambiguously define the stereochemistry. 8) It has been reported that the acetamido group in N-acetyl- α , β -dehydroalaninates exercises a strong directing effect in the addition of electrophilic reagents to the double bond. See, for example, A. L. Love and R. K. Olsen, J. Org. Chem., <u>37</u>, 3431 (1972). 9) The reactivity and the regio-selectivity in 1,3-dipolar cycloadditions have been explained cleanly by the application of Frontier Orbital Theory: I. Fleming, "Frontier Orbitals and Or-ganic Chemical Reactions", Wiley-Interscience, London, 1976 pp148-161. 10) R. Huisgen and H. U. Reissig, Angew. Chem. Int. Ed. Engl., <u>18</u>, 330 (1979). 11) T. Hiyama and M. Kai, Tetrahedron Lett., 2103 (1982).